

# On the Study of Racah Parameter B for Nickel (II) and Cobalt (II) Compounds

Mohamad M.E. Shakdofa<sup>1</sup>, Kamal A. Aly<sup>2,3</sup>, Pankaj Sharma<sup>4</sup> and Gomaa A. M. Ali<sup>5,\*</sup>

<sup>1</sup>Chemistry Department, Faculty of Science and Arts, P.O. Box 80200, 21589 Khulais, University of Jeddah, Jeddah, Saudi Arabia

<sup>2</sup>Physics Department, Faculty of Science and Arts, P.O. Box 80200, 21589 Khulais, University of Jeddah, Jeddah, Saudi Arabia

<sup>3</sup>Physics Department, Faculty of Science, Al-Azhar University, P.O. 71452, Assiut, Egypt

<sup>4</sup>Department of Physics and Materials Science, Jaypee University of Information Technology, Waknaghat, Solan, H.P. – 173234, India

<sup>5</sup>Chemistry Department, Faculty of Science, Al-Azhar University, Assiut branch, Assiut 71524, Egypt

**Abstract:** The ligand field strength (Dq) and interelectronic repulsion (B) have been calculated for nickel (II) and cobalt (II) compounds by Underhill and Billing [1] based on  $v_2$  and  $v_3$  values. They solved the quadratic equation  $340(Dq^2 - 18(v_2 + v_3) Dq + v_2 v_3) = 0$  in Dq for octahedral. Then, the B value was determined using the equation  $v_2 + v_3 - 30Dq - 15B = 0$ . But, the first equation has real and imaginary solutions and it's difficult to determine the imaginary one. In addition, in many cases there is no solution of this equation as well as in the case of CoO ( $d^7$  oct.). For octahedral, the above two equations can be considered as a system of two equations with two unknowns which can be solved if we have  $v_2$  and  $v_3$  values and vice versa. Similarly, for tetrahedral, the former two equations also can be solved for Dq and B. Using Newton–Raphson iterations, Dq and B values are exactly determined for octahedral and tetrahedral systems. The obtained values of Dq and B were compared with those previously reported.

**Keywords:** Racah parameter, Metallic composites, Electronic transition.

## INTRODUCTION

The spectra of nickel (II) [2-4] and cobalt (II) [5,6] complexes have been used to estimate Racah parameter (interelectronic repulsion parameter,  $B$ ). In octahedral nickel (II) complexes, only the bands of  $v_2(3A_{2g} - 3T_{1g}(F))$  and  $v_3(3A_{2g} - 3T_{1g}(P))$  are observed. The ligand field strength (Dq) and interelectronic repulsion (B) have been estimated by solving the secular equation [7]. Underhill and Billing [1] simplify the problem by solving a quadratic equation in Dq and B only. The equations for calculating Dq and B based on the  $v_2$  and  $v_3$  values for octahedral Ni(II) complexes  $d^8$  are [1]:

$$340Dq^2 - 18(v_2 + v_3)Dq + v_2v_3 = 0 \quad (1)$$

$$B = \frac{v_2 + v_3 - 30Dq}{15} \quad (2)$$

Solving the quadratic equation will give two solutions (real and imaginary). It is difficult to determine

the imaginary one. To overcome this problem Eq. (2) can be rewritten as:

$$-30Dq + (v_2 + v_3) - 15B = 0 \text{ or } v_2 + v_3 = 30Dq + 15B \quad (3)$$

Then, substituting the term  $v_2 + v_3$  into Eq. (1) gives:

$$-200 Dq^2 - 270BDq + v_2v_3 = 0 \quad (4)$$

These two Eqs. (3) and (4) are contained two unknowns *i.e.* these two equations can be solved for Dq and  $B$  if we have  $v_2$  and  $v_3$  values and vice-versa. Each of the two systems of the Eqs. (1) and (2) or Eqs. (3) and (4) can be solved for Dq and  $B$  for Ni(3Etpy)<sub>2</sub>Br<sub>2</sub> ( $d^8$ , octa.), NiMbCl<sub>2</sub> ( $d^8$ , octa.), CoO ( $d^7$ , octa.) using a rapidly-converging algorithm (Newton–Raphson iterations) [8,9]. The obtained values have been listed in Table 1 and are compared with those reported previously. It has been found that, in the case of CoO ( $d^7$  octa.), there is no solution of quadratic Eq. (1). The two solutions of Eq. (1) are given by:

$$Dq = \frac{18(v_2 + v_3) \pm \sqrt{(18(v_2 + v_3))^2 - 4(340)v_2v_3}}{2(340)} \quad (5)$$

Substituting  $v_2 = 16680 \text{ cm}^{-1}$  and  $v_3 = 18450 \text{ cm}^{-1}$  [10], Eq. (5) could be written as:

\*Address correspondence to this author at the Chemistry Department, Faculty of Science, Al-Azhar University, Assiut branch, Assiut 71524, Egypt; E-mail: pks\_phy@yahoo.co.in

$$Dq = \frac{632340 \pm \sqrt{3.99 \times 10^{11} - 4.185 \times 10^{11}}}{680} \quad (6)$$

The value under the square root is negative. Therefore there is no solution of Eq. (1) at these values of  $\nu_2$  and  $\nu_3$ . Thus a high doubt on how it has been solved.

Furthermore, the authors' values of  $Dq$  ( $887 \text{ cm}^{-1}$ ) and  $B$  ( $780 \text{ cm}^{-1}$ ) for CoO don't satisfy the Eqs. (1) and / or (3) as one can see Eq. (1) for CoO take the following form:

$$(340 \times (887)^2) - (18 \times (16680 + 18450) \times 887) + (16680 \times 18450) \neq 0$$

Similarly, Eq. (2) become:

$$(-30 \times 887) + (16680 + 18450) - (15 \times 780) \neq 0$$

On the other hand,  $Dq$  and  $B$  values ( $781 \text{ cm}^{-1}$  and  $835 \text{ cm}^{-1}$ ) for  $\text{Ni}(\text{3Etpy})_2\text{Br}_2$  ( $d^8$ , octa.) complex does not satisfy Eq. (3) consequently, Eqs. (1) or (4) also can't be satisfied. Although,  $B$  ( $152 \text{ cm}^{-1}$ ) and  $Dq$  ( $1123 \text{ cm}^{-1}$ ) values for  $\text{Ni}(\text{3Etpy})_2\text{Br}_2$  ( $d^8$ , octa.) are satisfied by the Eq. (1), (2) and (3) but these values are far away from those previously reported. Therefore, these values are incorrect. The exact values of  $Dq$  and  $B$  are given in Table 1. From Table 1, it is noted that  $Dq$  and  $B$  values

are very accurate and sensitive, therefore the two Eqs. (1) and (2) or (3) and (4) must be solved exactly.

On the other hand, the corresponding relations for ions with  $d^2$  or  $d^7$  configuration in an octahedral field or with a  $d^3$  or  $d^8$  configuration in a tetrahedral field [1]:

$$340Dq^2 + 18(\nu_3 - 2\nu_2)Dq + \nu_2^2 - \nu_2\nu_3 = 0 \quad (7)$$

$$B = \frac{\nu_3 - 2\nu_2 + 30Dq}{15} \quad (8)$$

Rewriting Eq. (8) in the form of:

$$30Dq + (\nu_3 - 2\nu_2) - 15B = 0 \quad \text{or} \quad (9)$$

$$\nu_3 - 2\nu_2 = 15B - 30Dq$$

Substituting the term  $\nu_3 - 2\nu_2$  into Eq. (7) gives:

$$-200Dq^2 + 270BDq + \nu_2^2 - \nu_2\nu_3 = 0 \quad (10)$$

Similarly, the two Eqs. (9) and (10) have been solved using Newton-Raphson iterations [8, 9] in  $Dq$  and  $B$  for  $\text{NiM}_5\text{Cl}_2$  ( $d^8$ , tetra.) and  $\text{Cs}_3[\text{CoCl}_2]\text{Cl}$  ( $d^7$ , tetra.). The obtained values are listed in Table 1. In comparison with previously reported results, it has been found that, the  $Dq$ ,  $B$ ,  $\nu_2$  and  $\nu_3$  values don't satisfy Eq. (9) where the term  $\nu_3 - 2\nu_2$  isn't equal to the term  $15B - 30Dq$  (Table 1). In the case of  $\text{Cs}_3[\text{CoCl}_2]\text{Cl}$

**Table 1: Calculated Values of  $Dq$  and  $B$  ( $\nu_2$ ,  $\nu_3$   $Dq$  and  $B$  in  $\text{cm}^{-1}$ )**

Compound	Ref.	$\nu_2$	$\nu_3$	$Dq$	$B$	$\nu_2 + \nu_3$	$30Dq + 15B$	$\nu_3 - 2\nu_2$	$15B - 30Dq$	$Dq$	$B$	Reported values	
												$Dq$	$B$
		Ref. 1				octa		tetra		This study ++		$Dq$	$B$
$\text{Ni}(\text{3Etpy})_2\text{Br}_2$ (octa.)	[1, 2]	12970	23000	781 1123	835 152	35970	35950 35970	-2940	-10905 -31410	781.262	835.472	789	830
$\text{NiM}_5\text{Cl}_2$ (octa.) *+	[1,3]	13150	23000	795 1118	819 173	36150	36140 36140	-3300	-11565 -30945	795.29	819.42	794	825
CoO (octa.)	[1, 10]	16680	18450	887-98	780 ---	35130	38310 20790	- 14910	-14910 ---	884.298	781.403	888	780
2.5-DmpNiC1 <sub>2</sub> (octa.) *+	[3]	13000	22750	-----	1190 ---	35750	---	-3250	---	786.088	811.156	786	811
$\text{NiM}_5\text{Cl}_2$ (tetra.)	[1,3]	9200	15620	497-350	--- 808	24820	---	-2780	--- 22620	496.829	808.325	500	810
$\text{Cs}_3[\text{CoCl}_4]\text{Cl}$ (tetra.)	[1,6]	5580	14800	320 759	-884 719	20380	---	3640	-22860 ---	304.397	851.46	320	719
$\text{M}_5\text{NiBr}_2$ (tetra)	[3]	8600	15470	---	-159 ---	24070	---	-1730	---	465.16	814.987	465	820
$\text{M}_5\text{NiI}_2$ (tetra)	[3]	8100	15400	---	---	23500	---	-800	---	438.741	824.149	435	825

3Etpy is 3-ethyl pyridine, Mp is Methyl pyrazine, Dmp is Dimethyl pyrazine, \* Bridging Halogen, + Bridging pyrazine, ++ These values satisfy that the term  $\nu_2 + \nu_3 = 30Dq + 15B$  for octahedral and the term  $\nu_2 - 2\nu_3 = 15B$  for tetrahedral.

( $d^7$ , tetra.), solving the Eqs. (9) and (10) gives Dq and B values as  $304.397 \text{ cm}^{-1}$  and  $851.46 \text{ cm}^{-1}$ . These values of Dq and B are in excellent agreement when we solve the quadratic Eq. (7) in Dq according to the Ref [1] authors. The solutions for Dq are  $304.397 \text{ cm}^{-1}$  and  $-497.1 \text{ cm}^{-1}$ . Substituting Dq values into eq. (8) gives B values of  $851.46 \text{ cm}^{-1}$  and  $-751.54 \text{ cm}^{-1}$  while, the authors of the Ref. [1] reported that Dq =  $320 \text{ cm}^{-1}$  and B =  $719 \text{ cm}^{-1}$ . In the case of NiM<sub>2</sub>Cl<sub>2</sub> ( $d^8$ , tetra.) a small difference has been observed, but the exact values that satisfies the system of Eqs. (9) and (10) are needed.

For clarity and confirmation Dq and B values for octa 2.5-DmpNiC<sub>12</sub>, tetra Mp<sub>5</sub>NiBr<sub>2</sub> and tetra Mp<sub>5</sub>NiI<sub>2</sub> have been calculated based on  $\nu_2$  and  $\nu_3$  values from Ref [3] (see Table 1) and also compared with the reported data [3]. Small difference is observed between the reported values and the present study. Furthermore, the exact values of Dq and B according to the present study satisfy the pairs Eqs. (1) and (2) or/and (3) and (4) for ions with octahedral field and also satisfy the pairs Eqs. (7) and (8) and/or (9) and (10) for ions with tetrahedral field.

## HIGHLIGHTS

- In the previous reports Dq and B for oct. and tet. compounds were determined by solving the secular equation only.
- The secular equation has real and imagery solutions, it's difficult to determine the imaginary one.

- The present study concerns with precisely determination of both Dq and B using Newton–Raphson iterations.

## REFERENCES

- [1] Underhill AE and Billing DE. Calculations of the Racah Parameter B for Nickel (II) and Cobalt (II) Compounds, Nature 1966; 210: 834-835.  
<https://doi.org/10.1038/210834a0>
- [2] Lever ABP and Nelson SM. An Analysis of the Electronic Spectra of Bisarninecobalt Halides. The Effect of Low Symmetry and of Steric Hindrance. Chem Commun 1965; 9: 168-169.  
<https://doi.org/10.1039/c19650000168>
- [3] Lever ABP, Lewis J and Nyholm RS. A Spectroscopic Xstudy of Some Pyraxine-NicEel(II) Complexes. J Chem Soc 1964; 0: 4761-4768.  
<https://doi.org/10.1039/JR9640004761>
- [4] Goodgame DML, Goodgame M and Weeks MJ. Spectral and magnetic studies of some polymeric complexes of nickel halides with heterocyclic ligands. J Chem Soc 1964; 0(0): 5194-5199.  
<https://doi.org/10.1039/jr9640005194>
- [5] Lever ABP, Nelson SM and Shepherd TM. Calculations of the Racah Parameter B for Nickel (II) and Cobalt (II) Compounds. Inorg Chem 1965; 4: 810-813.  
<https://doi.org/10.1021/ic50028a008>
- [6] Cotton FA, Goodgame DML and Goodgame M. The Electronic Structures of Tetrahedral Cobalt (II) Complexes. J Amer Chem Soc 1961; 83(23): 4960-4699.  
<https://doi.org/10.1021/ja01484a002>
- [7] Jorgensen CK. Absorption Spectra and Chemical Bonding in Complexes 1962; London: Pergamon Press.
- [8] Aly KA. Optical constants of quaternary Ge-As-Te-In amorphous thin films evaluated from their reflectance spectra, Philosophical Magazine 2009; 89(12): 1063-1079.  
<https://doi.org/10.1080/14786430902870542>
- [9] Allgower E and Georg K. Numerical continuation methods: an introduction. Springer Series in Computational Mathematics. 1990, New York: Springer Berlin Heidelberg.  
<https://doi.org/10.1007/978-3-642-61257-2>

Received on 13-12-2017

Accepted on 28-12-2017

Published on 31-12-2017

DOI: <http://dx.doi.org/10.15377/2409-5826.2017.04.5>

© 2017 Shakdofa, et al.; Avanti Publishers.

This is an open access article licensed under the terms of the Creative Commons Attribution Non-Commercial License (<http://creativecommons.org/licenses/by-nc/3.0/>) which permits unrestricted, non-commercial use, distribution and reproduction in any medium, provided the work is properly cited.